

Biosynthesis of copper nanoparticles using aqueous extract of *Capparis spinosa* fruit and investigation of its antibacterial activity

Katrin EBRAHIMI, Sima SHIRAVAND, Hossein MAHMOUDVAND

ABSTRACT

The present study was aimed to use the aqueous extract of *Capparis spinosa* to synthesize the copper nanoparticles and also evaluated their antibacterial activities against some pathogenic bacterial strains. UV-vis spectroscopy analyses, Fourier transform of infrared (FTIR), scanning electron microscopy (SEM), and energy dispersive X-ray (EDX) were used to identify the synthesized nanoparticles. The antimicrobial activity of the synthesized copper nanoparticles was investigated using disk diffusion method and broth microdilution against some Gram-positive and Gram-negative bacteria. After adding the extract to the copper sulfate solution, the color of the solution changed from light blue to yellowish green. Existence of a maximum peak at the wavelength of 414 nm confirmed the

formation of the copper nanoparticles. FTIR spectrum analysis showed that the factor groups created a coating extract on the surface of the nanoparticles. Scanning electron microscopy demonstrated the particle size between 17 and 41 nm. These findings showed that *Staphylococcus aureus* and *Bacillus cereus* as Gram-positive bacteria were most susceptible to synthesized copper nanoparticles in comparison with the Gram-negative bacteria (*Klebsiella pneumoniae*, and *Escherichia coli*). The obtained findings demonstrated that the aqueous extract of *C. spinosa* acts as a reducer and stabilizer factor. The synthesized copper nanoparticles demonstrated activity against both Gram-positive and Gram-negative bacteria.

Keywords: Nanoparticles; antimicrobial; green synthesis; copper

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1. Introduction

Nanotechnology plays an important role in the modern research [1]. This technology can be applicable in a wide range of fields such as all pharmacology, food and nutrition, chemical industries, energy sciences, cosmetics; further, it can be used for the treatment of infections, cancer, allergies, diabetes, and inflammation [1-5]. Green chemistry refers to the design and development of chemical products and processes in order to minimize the dangerous uses of the environment [6]. Nanoparticles are synthesized through chemical and physical methods. Compared to these methods, green synthesis has been one of the best methods for producing nanoparticles in recent years. Green synthesis has numerous advantages, compared to other methods, including cost-effectiveness, simplicity, use of lower temperatures, use of non-toxic materials, as well as compatibility with applied medical and nutritional programs [7]. Green synthesis method is being developed and is an environmentally-friendly method as well [7-8]. In this method, the extract is used as a reducing agent and coating for nanoparticles. Owing to their electrical, optical, and catalytic properties, copper nanoparticles are widely used and have

various medical, antifungal, and antibacterial applications [9, 10]. Copper nanoparticles are toxic for many microorganisms such as *Escherichia coli*, *Staphylococcus aureus*, and *Pseudomonas aeruginosa* bacteria and, in the meantime, are non-toxic for animal cells [11]. Several different plants have been used for the synthesis of nanoparticles using green synthesis method [12]. Nanoparticles are synthesized using all parts of the plant including stem, flower, fruit, leaves, and bark [13]. Cappariaceae is a large family of the flowering, angiospermous, dicotyledonous, and dialypetalous plants and *Capparis* is the largest genus in this family [14]. The scientific name of *Capparis spinosa* is known as Koor, Kabar, Kooraz, Koorgiah, Khiarshang, Alafmar, Lijin, and Daghgarpooziin Iran; further, it is called Batlisk in Lorestan Province, western Iran [14]. This plant is a shrub with stalks lying on the ground, simple leaves with thorn-like stipules, large and fragrant white or purple flowers, and unequal convex sepals with 8 to 15 stamens [15]. In the present study, the fruits of this plant were used for the synthesis of copper nanoparticles and investigation of their antibacterial effects.

2. Results

After adding the extract to the copper sulfate solution, the color of the copper sulfate changed from light blue to yellowish green, which was due to the synthesis of the copper nanoparticles.

2.1. UV-Vis spectrum analysis

The nanoparticles synthesized in the zone of 414 nm had a peak maximum. The results show that the characteristic of the resonance band of the surface plasmon at the wavelength of 414 nm occurred for the copper nanoparticles.

2.2. FTIR spectrum analysis

Results of FTIR showed that the biomolecules in the extract reduced the copper sulfate solution; further, they would be used as coatings for nanoparticles. The bands at 3380, 2928, 1741, 1604, 1400, 1050, and 1271 were related to the O-H stretching of alcohol and phenol, C-H stretching of aliphatic group, C=O stretching of ester carbonyl, C=C stretching of the aromatic ring, and C-O stretching of ester, respectively.

The bands showed that the extract of *C. spinosa* was rich in polyphenols, sterols, and sorptive fatty acids in the zones of 3398, 2924, and 1739, respectively. Since the copper nanoparticles had bands at 1624, 1359, and 1072, it could be concluded that the biomolecules of the extract acted as both reducer and coating for copper nanoparticles; thus, they can protect the copper nanoparticles against oxidation and being transformed into copper oxide.

2.3. Scanning electron microscope (SEM)

After confirming the synthesis of nanoparticles using color change and Vis-UV and FTIR sorption spectra, the morphology of the synthesized nanoparticles was investigated by the scanning electron microscope at University of Lorestan, Khorramabad, Iran. Based on the obtained images, the synthesized copper nanoparticles had spherical morphology and the size of the particles was determined between 17 and 41 nm (Figure 1).

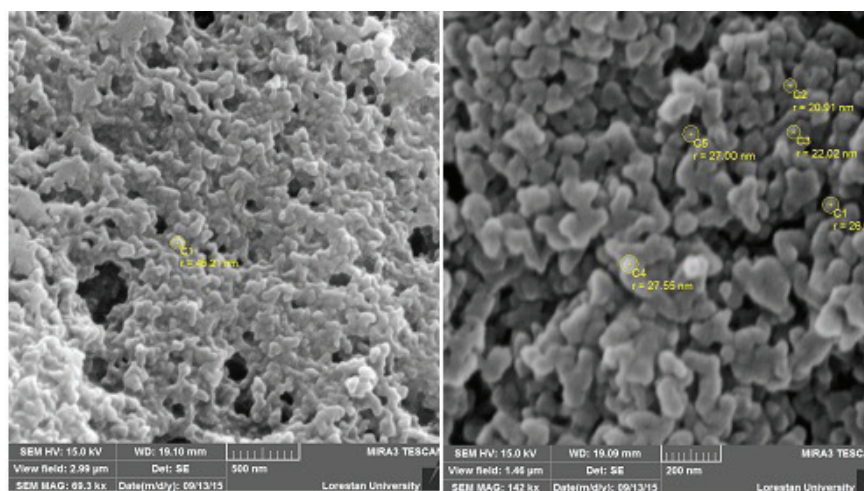


Figure 1. Scanning electron microscope of copper nanoparticles synthesized using aqueous extract of *Capparis spinosa* fruit

2.4. Investigating EDX spectrum

The presence of metallic copper was confirmed by EDX analysis. The copper nanoparticles had a sorptive peak at 1Kev, which is an index for metallic nanoparticles of copper. Oxygen was the main contamination in this spectrum. Oxygen was resulted from physical absorption during sample preparation.

2.5. Antibacterial activity of synthesized nanoparticles

Table 1 shows the antibacterial effects of synthesized copper nanoparticles against some pathogenic bacterial strains. Copper nanoparticles demonstrated remarkable antibacterial activity against Gram-positive bacteria (*S. aureus* and *B. cereus*) and Gram-negative bacteria (*K. pneumoniae* and *E. coli*). These findings showed that *S. aureus* and *B. cereus* as Gram-positive bacteria were most susceptible to synthesized copper nanoparticles in comparison with the Gram-negative bacteria.

Table 1. Minimum inhibitory concentration (MIC) and minimum bactericidal concentration (MBC) values copper nanoparticles against some pathogenic bacterial strains.

| Bacterial strain | MBC(mg/ml) | MIC (mg/ml) |
|------------------------------|------------|-------------|
| <i>Bacillus cereus</i> | 5 | 5 |
| <i>Staphylococcus aureus</i> | 5 | 5 |
| <i>Escherichia coli</i> | 10 | 10 |
| <i>Klebsiella pneumoniae</i> | 10 | 10 |

3. Discussion

Investigating the results showed that the use of the aqueous extract of *C. spinosa* is a simple, quick, inexpensive, environmentally-friendly, and applicable method which can be performed in any kind of laboratory for the synthesis of copper nanoparticles. In this method, the chemical and toxic reagents are not used and no pollution is created for the environment; therefore, it is superior to other physical and chemical methods of nanoparticle synthesis. Nanoparticles produced by this method are very stable and renewable. Copper nanoparticles have various applications in medicine, including anti-cancer, anti-parasitic, antifungal, and antibacterial activities, food packaging, and wound dressings; moreover, these nanoparticles have various industrial applications including m electrical capacitors, heat transfer, super-strong materials, sensors, catalysts, etc [2]. Results showed that the nanoparticles synthesized by this

method were nearly spherical with the size ranging from 41 to 17 nm; further, the synthesized copper nanoparticles had the capability of fighting the pathogenic microorganisms. The observed growth inhibition halo against the Gram-positive bacteria was more than the halo against the Gram-negative ones. So far, different plants have been used for the synthesis of copper nanoparticles, including *Ocimum sanctum* by Kulkarni (2014), *Capparis zeylanica* by Saranya Adevi et al. (2014), *Gloriosa superbal* by Naika et al. (2015), extract of *Vitis vinifera* by Subbaiya Angrasan (2014), *Nerium oleander* by Gopinath et al. (2014), and *Artabotrys odoratissimus* by Kathad and Gajera (2014) [16-21].

EDX images proved the existence of copper nanoparticles and, using SEM, their size was determined to be less than 50 nm. Saranya Adevia et al [17]. expressed the size of the synthesized particles between 100 and 50 nm.

Subhankari et al. reported the particle size of 25-40 nm [22]. Kulkarni reported the particle size of 77 nm, while Shend et al. reported the particle size of 60-10 nm [23]. Investigating the antibacterial activity of the nanoparticles against Gram-positive and Gram-negative bacteria showed that the effect of these particles on the Gram-positive bacteria was greater than the one on the Gram-negative bacteria. Results of this study were consistent with the results presented by Angrasan and Subbaiya.

Conclusion

Results showed that the aqueous extract of *C. spinosa* acts as a reviver and stabilizer factor. The synthesized copper nanoparticles demonstrated activity against both Gram-positive and Gram-negative bacteria.

4. Materials and methods

4.1. Plant material

The *Capparis spinosa* fruits were collected from rural regions of Kouhdasht, Lorestan Province, in July 2016 by Mr. Mohammad Mehrnia, and identified by Ms. Katrin Ebrahimi. A voucher specimen was deposited in the Herbarium of Agricultural Research Center, Khorramabad, Iran (No. 13840).

Fruits of *C. spinosa* were collected from city of Kouhdasht, Lorestan Province. After being identified, the collected specimens were washed with distilled water and dried in the shade away from the direct sunlight; then, they were milled to powder. Afterwards, the powder of the plant was stored in the refrigerator for later uses.

4.2. Preparation of aqueous extract

First, 10 g of the *C. spinosa* powder was poured in a flask and 200ml of deionized water was added to it; then, it was heated for 30 min at 70°C. After reaching the room temperature, it was first filtrated with a filter paper and, then, the extract was centrifuged for 20 min at the speed of 12000 rpm. Next, the extract was stored in the refrigerator at 4°C [24, 25].

4.3. Synthesis of copper nanoparticles

First, 75 ml of the freshly prepared extract was added to 100 ml of the freshly prepared 0.01 M copper sulfate solution while being constantly stirred on the stirred and then it was stored at 60°C for 24 h [26]; afterwards, it was centrifuged for 20 min at the speed of 12,000 rpm. This procedure was repeated twice in order to remove all the impurities. The color change of the solution from green to amber yellow indicated the formation of the nanoparticles. The synthesized nanoparticles were dried in the oven at 60°C for the following analyses [27, 28].

4.4. Detection of nanoparticles

4.4.1. UV-Vis spectroscopy analysis

Reduction of the copper ions to copper nanoparticles is the confirmation of surface plasmon resonance (SPR) of the copper nanoparticles. Thus, 300 µl of the sample was diluted with 3 ml of distilled water, and UV-Vis spectrum analysis was performed using a spectrophotometer device (JENWAY 6405) in the range of 300-700 nm [29, 30].

4.4.2. Fourier transform infrared spectroscopy

FTIR spectrum analysis for *C. spinosa* extract and copper nanoparticles synthesized using the extract indicated the presence of the biological agents in the extract and led to the reduction of Cu^{2+} to copper nanoparticles. The specimen and KBr granules were powdered together with the ratio of 1 to 100 (1/100 ratio) and then compressed into tablets; subsequently, the analysis was performed and measured using FTIR spectrophotometer (model Nicolet32) in the range of 400-4000 and with the resolution of 1-4cm [31].

4.4.3. Scanning electron microscope (SEM)

Size and morphology of the synthesized nanoparticles were examined using electron microscopy (Mira3, Made in Czech) with 15 kv, magnification of x10, and resolution of 1 nm.

4.4.4. Energy dispersive X-ray (EDX)

Energy dispersive X-ray spectroscopy along with SEM was used to investigate the presence of copper in SEM images.

4.5. Preparation of culture medium and bacterium

The solid culture and liquid culture media are used for the disk diffusion and MIC methods, respectively; therefore, in the present study, the Mueller Hinton agar and the Mueller Hinton Broth media were used as the solid and liquid culture media, respectively. To prepare the culture medium according to the written instruction, on the culture medium plate, a specified amount of the culture medium was dissolved in a specified volume of distilled water in an erlenmeyer flask. Then, it was placed in an autoclave to sterilize the culture medium. After the autoclave process, the liquid culture medium was put in the refrigerator. After reaching the temperature of 50°C, the culture medium was poured in multiple plates under the hood; afterwards, when the culture medium turned into the solid form, the plates were inverted upside down and then stored in the refrigerator. During the test, the newly passaged bacteria should be used; for this purpose, a fresh culture was prepared from each bacterium just one day before the test, which was done with a loop on the Mueller Hinton agar culture medium. Plates were put in the incubator for 24 h.

4.6. Disk diffusion measurement

The disk diffusion method was used to investigate the antibacterial activity of the nanoparticles against *S. aureus* (PTCC 1112), *B. cereus* (PTCC 1556), *E. coli* (PTCC 1330), and *K. pneumoniae* (PTCC 1053) [32]. Under sterile conditions, a loop of the 24 h culture of each bacterium was poured in the normal saline solution to yield the same opacity as 0.5 McFarland standard (1.5×10^6 CFU/ml); then, a homogeneous culture was prepared by a swap on the surface of the plate containing the Mueller Hinton agar culture medium. Next, 100 µl of the specimen was poured on a blank disc, and the blank disc was fixed on the surface of the plate. Normal saline and amikacin were used as the negative and positive controls, respectively. After 24 h of storage at 37°C in the incubator, the diameter of the inhibition halo was measured.

4.7. Determination of MIC

MIC was determined in a sterile 96-well microplate using broth microdilution method [33]. For this purpose, first, 100

µl of the Mueller Hinton broth culture medium (MHB) was poured in 10 wells of the microplate and 100 µl of the specimen was added to the first well of each row and thoroughly mixed with the sampler by turning up and down multiple times. Afterwards, 100 µl of the content of the first well was taken out and then added to the next well. This procedure was repeated until the 10th well. Subsequently, 100 µl was poured out from the 10th well so that the concentration of the extract in each well became half of the previous well. From the 24 h the bacterium culture, a homogeneous suspension equivalent to the 0.5 McFarland standard solution was prepared in the broth liquid culture medium and diluted 100 times; then, 100 µl was added to each well. After 24 h of incubation at 37°C, the existence of opacity indicating the growth or non-growth of the bacterium was recorded; then, based on the definition, the most diluted well with no opacity was considered as MIC. The bacterial suspension and culture medium, the MHB, and the amikacin were used as the positive control, negative control, and standard antibiotic, respectively.

4.8. Statistical analyses

We used SPSS software, ver. 17, (SPSS Inc., Chicago). Data entry and statistical analysis and differences between the groups were determined using one-way analysis of variance (ANOVA) test. P value of less than 0.05 was considered statistically significant.

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Declaration of Interest

The authors declare that there is no conflict of interest in this study.

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