RESEARCH PAPER

HPLC Assay Method Development and Validation for Quantification of Capecitabine in Tablets and Forced Degradation Samples

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ABSTRACT

A simple, rapid, accurate and stability indicating RP-HPLC method was developed for the determination of Capecitabine in pure and tablet dosage form. The mobile phase consisting of 0.1% aqcetic acid, methanol and acetonitrile in the ratio of 35:60:5 ν/ν . The Inertsil ODS (octadecyl silane), C18, 3V, 250 x 4.6 mm, 5µm with UV detection 304 nm was used. The retention time was found to be 6.4 minutes. The method was statistically validated for accuracy, linearity, precision, robustness, specificity and range. The method was found linear over the concentration range of 50-150 µg/ml. The recovery studies of dosage form were also carried out and analyzed; the % relative standard

deviation (RSD) from recovery studies was found to be within limits. The specificity of the method was ascertained by forced degradation studies by acid, alkali hydrolysis and oxidation. About 35% of drug is degraded in acidic medium and less than that in alkaline and oxidative conditions. Due to simplicity, rapidity and accuracy of the method, the method will be useful for routine analysis and checking purity of capecitabine tablet. The method further can be investigated for pharmacokinetic and biopharmaceutical analysis.

Keywords: Capecitabine, 5-fluorouracil, metastatic breast, colorectal cancer, RP-HPLC, ICH guidelines.

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INTRODUCTION

Capecitabine has the chemical name pentyl [1-(3,4-dihydroxy-5-methyl-tetrahydrofuran-2-yl)-5-fluoro-2-oxo-1Hpyrimidin-4-yl]aminomethanoate [1]. It is a white to almostwhite solid with a molecular formula of C₁₅H₂₂FN₃O₆ and a molecular weight of 359.3 (Fig. 1). Capecitabine tablets contain NLT 93.0% and NMT 105.0% of the labeled amount of Capecitabine [2]. Capecitabine is an orally-administered chemotherapeutic agent used in the treatment of metastatic breast and colorectal cancers. Capecitabine is a prodrug, that is enzymatically converted to 5-fluorouracil in the tumor, where it inhibits DNA synthesis and slows growth of tumor tissue. The activation of Capecitabine follows a pathway with three enzymatic steps and two intermediary metabolites, 5'-deoxy-5-fluorocytidine (5'-DFCR) and 5'-deoxy-5fluorouridine (5'-DFUR), to form 5-fluorouracil [3]. The final step is the tumor-activated conversion to 5-FU by TP, predominantly in malignant cells. This site-specific action may result in higher tumor site concentrations and lower systemic toxicity than that seen with intravenous use of 5-FU [4].

Capecitabine is administered orally. Absorption is rapid with peak plasma levels occurring in about 1.5 hours (compared to 2 hours for 5-FU). Food decreases the rate of absorption of Capecitabine (60% reduction in peak plasma concentration) and AUC by 35%. Less than 60% of Capecitabine and its metabolites are bound to plasma protein. The elimination half-life of both Capecitabine and 5-FU is about 0.75 hours, with more than 70% of the administered dose recovered in urine, largely as the alanine metabolite of 5-FU [5]. The absolute bioavailability of Capecitabine is 42% with low interpatient variability [6].

The usual starting dose is $2,500 \text{ mg/m}^2/\text{day}$ in two divided doses, 12 hours apart. One cycle includes two weeks of treatment followed by one week without treatment. Cycles can be repeated every three weeks [7].

There are many methods which were developed and validated for the determination of capecitabine from various formulations. In that the HPLC-UV, liquid chromatography - tandem mass spectroscopy and automatic pressure chemical ionization LC-MS/MS method for determination of capecitabine and its metabolites from human plasma, using C₁₈ reversed phase column with formic acid solution (pH-3): ethanol (55:45, v/v) as mobile phase, Atlantis dC18 column under gradient elution respectively which were validated according to the FDA guidelines [8-15]. Another liquid chromatography- tandem mass spectrometry method for determination of capecitabine in dried blood spot, separation was done on Phenomenex Gemini C₁₀ column, isocratic condition for mobile phase as acetonitrile : 2 mmole ammonium formate (80:20, v/v) [16]. Reversed phase method was developed for capecitabine determination in bulk and pharmaceutical formulations where separation performed on Welchrom C₁₈ column with isocratic mobile phase containing methanol: acetonitrile: water $(50:30:20, \nu/\nu)$ which was monitored at 245 nm, method was validated according to ICH guidelines [17]. Spectrophotometric methods were developed for determination of capecitabine based on their oxidation and precipitation reactions, which was monitored

Table 1. Chromatographic conditions

by measuring absorbance of produced complexes [18]. HPLC method was developed for the simultaneous determination of capecitabine and its metabolites in mouse plasma, liver, human xenograft tumors with electrospray ionization and detected by mass spectrometry, this method was cross validated in human plasma and human tumor for clinical applications [19]. Hence there is need to develop simple, accurate, sensitive and selective method for quantification of capecitabine.

MATERIALS AND METHODS

Chemicals and reagents

A HPLC grade, acetic acid, acetonitrile, and methanol were procured from Merck Labs. Mumbai. Hydrochloric acid, sodium hydroxide and hydrogen peroxide were obtained from Merck Labs. Mumbai. Distilled water was obtained from Milli Q water system in laboratory. Capecitabine reference standard was obtained from Glenmark Pharmaceuticals Ltd., Sinnar. The marketed formulation of capecitabine was procured from local market.

Instrumentation and chromatographic conditions

An isocratic HPLC system of Schimadzu LC2010 UV detector having software LC solution was used for chromatographic analysis. The chromatographic conditions were optimized and these were as shown in Table 1.

Solutions Preparation

Preparation of Mobile Phase

A 0.1 % solution of acetic acid is prepared by dissolving 1 ml of glacial acetic acid in sufficient distilled water to produce 1000 ml. A mixture of 0.1% acetic acid, methanol and acetonitrile was prepared in the ratio of 35:60:5, v/v/v and degassed by sonication for 30 minutes.

Column	Inertsil ODS, 3V, 250 x 4.6 mm, 5 μm, C18
Mobile Phase	0.1% acetic acid, methanol and acetonitrile 35:60:5, $\nu/\nu/\nu$
Detector (Wavelength)	UV 250 nm
Flow rate	1.0 ml/min.
Column temperature	40°C
Injection volume	10 µl
Run time	10 minute

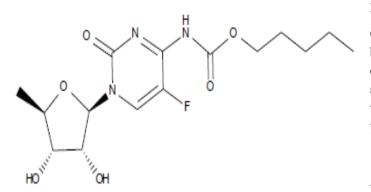


Figure 1. Structure of Capecitabine

Preparation of Standard Stock Solution

A 60 mg of capecitabine pure drug was dissolved in 60 ml of mobile phase. It was then sonicated for 5 minutes and made up the volume to 100 ml. This solution is then degassed using sonicator for 30 minutes ($600 \mu g/ml$).

Preparation of Sample

Twenty tablets were weighted. Average weight of tablets was calculated. All tablets were crushed to powder form. A quantity of powder containing about 60 mg of capecitabine was weighed accurately and transferred into 100 ml volumetric flask. About 70 ml of mobile phase was added and sonicated for 15 minutes. It was then cooled to room temperature and volume was made up to 100 ml. It was then passed through PVDF $0.45\mu m$ filter, and used thereafter (600 $\mu g/ml$). Appropriate aliquot was taken, diluted using mobile phase and analyzed by the proposed method.

Table 2. Assay and precision study of capecitabine tablet

Development of Calibration Curves/ Linearity

Calibration curve for capecitabine standard drug was plotted by preparing sample solution by diluting standard solution of capecitabine in mobile phase. From this stock solution, solutions of 50, 70, 90, 110, 130 and 150 μ g/ml concentration were prepared. Analysis of these solutions was carried out using chromatographic conditions, specified in Table 1.

Assay of Tablet

From the triturate of 20 tablets, an amount equivalent to 60 mg of capecitabine was weighed and dissolved in 60 ml of mobile phase. It was then sonicated for 10 minutes. The solution was filtered through 0.45μ PVDF syringe filter and then final volume of the solution was made up to 100 ml with mobile phase. Appropriate aliquot was taken, diluted using mobile phase and analyzed by the proposed method. Capecitabine tablets contain NLT 97.0 % and NMT 102.0 % of the labeled amount of capecitabine (C₁₅H₂₂FN₃O₆) (Table 2).

Method Validation

Methods validation is the process of demonstrating that analytical procedures are suitable for their intended use. Following aspects were covered during validation of developed method as per ICH guidelines [17].

Accuracy

Accuracy expresses the closeness of agreement between theoretical and practical value. The average result of mean for each level should be within 98 -102% and relative standard variation should not be more than 2%. The accuracy study

Sr. no.	Sample	Inter day		Intra day	
		Area	Capecitabine (%)	Area	Capecitabine (%)
1.	Precision Set 1	9435871	96.9	9440865	97.03
2.	Precision Set 2	9478159	98.3	9450325	97.6
3.	Precision Set 3	9460601	98.0	9487421	98.6
4.	Precision Set 4	9510168	97.6	9450627	97.3
5.	Precision Set 5	9457224	97.8	9457324	97.8
6.	Precision Set 6	9434868	97.5	9506984	97.4
		Average	97.68	Average	97.87
		Std. Dev.	0.48	Std. Dev.	0.47
		% RSD	0.49	% RSD	0.48

Std. Dev. is standard deviation and % RSD is % relative standard deviation.

Sr. no.	Sample Identity	Amt. added in mg	Avg. Area	Amount Recovered (mg)	Recovery (%)
1.	Accuracy-50%	30.62	5208950	30.71	101.22
2.	Accuracy-100%	61.02	10326070	60.89	100.69
3.	Accuracy-150%	87.16	14365781	84.70	98.07
				Average Std. Dev.	99.99 1.69
				% RSD	1.69

Table 3. Accuracy study of capecitabine tablet

Amt. is amount, Avg. is average, Std. Dev. is standard deviation and % RSD is % relative standard deviation. For average area (n-=3).

was carried out by recovery study. The recovery study was carried out by spiking pure capecitabine in marketed formulation sample for analysis and determining amount of pure drug recovered from same. The results of accuracy study are reported in Table 3.

Precision

The precision reflects closeness of agreement between a series of results determined by analyzing multiple samples of single analyte. According to ICH guidelines, precision can be determined by repeatability and intermediate precision study. Repeatability expresses degree of obtaining same result after analysis of sample over a short period of time. Intermediate precision deals with study of within-laboratory variations like different days. The results of precision study are reported in Table 2.

Specificity

Specificity reflects ability to quantify unequivocally the analyte in the presence of components like impurities,

degradants, and matrix, which may be expected to be present. Influence of such interfering components on retention time of capecitabine can be assessed by forced degradation studies. These were carried out by analysis sample after exposing it to 0.1 M hydrochloric acid (acidic), 0.1 M sodium hydroxide (basic) and 10% hydrogen peroxide (oxidative) solution. The results of specificity study are reported in Table 4.

Robustness

Robustness indicates capacity of method to remain unaffected due to small but deliberate variations in method parameters like flow rate, wavelength of analysis, and or temperature of analysis etc. The results of robustness study are reported in Table 5.

Linearity and Range

Linearity indicates obtaining test results which were directly proportional to the concentration of analyte in the sample within range specified. Solutions of 10, 30, 50, 70, 90, 110, 130, 150, 170, 190 and 210 μ g/ml concentration were injected.

Table 4.	Specificity	study of	capecitabine tablet
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Sr. no.	Sample	Name	Area	Capecitabine (%)
1.	Control	Capecitabine	9866577	101.4
2.	0.1M HCl	Degradants	2184640	32.5
		Capecitabine	6503440	67.5
3.	0.1M NaOH	Degradants	962765	14.6
		Capecitabine	8245257	85.4
4.	10% H ₂ 0 ₂	Degradants	408846	7.16
	2 2	Capecitabine	9072404	93.1

HCl is hydrochloric acid, NaOH is sodium hydroxide, H₂O₂ is hydrogen peroxide and 0.1 M means 0.1 molar.

The method was found linear in concentration range of 50 to 150μ g/ml.

The range is an interval between upper and lower concentration of the analyte in the sample where method was found linear. Linearity and range study was carried out together. The results of linearity and range study are reported in Table 6.

System Suitability Testing

System suitability testing is an integral part of many analytical procedures. The tests are based on the concept that the equipment, electronics, analytical operations and samples to be analyzed constitute an integral system that can be evaluated as such. System suitability test parameters to be established for a procedure depend on the type of procedure being validated. The results of system suitability studies are reported in Table 7.

RESULT AND DISCUSSION

Capecitabine is an effective anti-cancer agent used in the treatment of metastatic breast and colorectal cancers and is a prodrug, that is enzymatically converted to 5-fluorouracil. The quantitative estimation of capecitabine is possible by spectroscopic and chromatographic methods as reported. But most of method required buffer solution for chromatographic elution [12]. Whereas few method made use of gradient elution for same [10-11]. Hence it becomes necessary to develop simple chromatographic method with isocratic elution, without buffer solution as mobile phase component. In this method, it has been achieved by developing and validating simple, accurate, precise, sensitive and selective method for quantification of capecitabine.

The chromatographic conditions were optimized on trial and error method. The wavelength of analysis as 304 nm was finalized by analyzing 10 μ g/ml solution of capecitabine on UV-Visible spectrophotometer. The capecitabine is an

Table 5. I	Robustness	study	of cap	ecitabine	tablet
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Sr. No.	Parameter		Area		Avg. Capecitabine (%)	Std. Dev.	% RSD
		Set I	Set II	Set III			
1.	High flow (1.1 ml/min)	9128302	9115255	9083072	98.20	0.26	0.26
2.	Low flow (0.9 ml/min)	10821038	10817468	10825702	98.23	0.18	0.19
3.	High temp. (45°C)	9714497	9712398	9702982	99.24	0.41	0.41
4.	Low temp. $(35^{\circ}C)$	9699920	9710011	9693852	99.38	0.66	0.66

Std. Dev. is standard deviation and % RSD is % relative standard deviation. For Avg. of Capecitabine % (n=3).

Sr. no.	Concentration (%)	Area
1.	50	5304893
2.	70	83005483
3.	90	9235318
4.	110	10155951
5.	130	11321934
6.	150	14950420
9	Slope	16227
1	ntercept	527842
(Correlation coefficient	0.99958

Table 6. Linearity and range study of capecitabine tablet

Table 7. System suitability parameters

Parameters	Limits (Units)
Tailing factor	< 1.25
Therotical plates	> 4000
Retention time	6.4 minute
Standard deviation	< 1.0 %
% RSD	< 2.0 %
Correlation coefficient	<0.9977

Table 8. Summary of results

Parameter	Acceptance criteria (Units)	Result obtained (Units)
Precision	% RSD < 1.5%	0.49 %
Accuracy	% RSD <2%	1.69 %
Linearity	Correlation coefficient >0.9999	0.99958
Robustness: High flow		0.26 %
Robustness: Low flow	% RSD < 1.5%	0.19 %
Robustness: High pH		0.41 %
Robustness: Low pH		0.66 %
Specificity	Forced Degradation	In 0.1M HCl 32.5% degraded
Range	Normally accepted range 70-130 %	90-110 %

organic entity hence it will have assumed that it will be eluted on bonded stationary phases like C8 or C18. Hence, reverse phase chromatographic analysis has been proposed. The it has been observed that capecitabine was eluted on Inertsil ODS, (250 x 4.6 mm, 5 μ m) C18 column with good retention parameters. The selectin of mobile phase was carried out by changing composition of methanol and acetonitrile over wide range. But tailing of capecitabine peak was observed. Hence 0.1% acetic acid solution was used along with methanol and acetonitrile. Finally, capecitabine was eluted with good retention parameters using 0.1% acetic acid, methanol and acetonitrile (5:60:5, ν/ν), Inertsil ODS, (250 x 4.6 mm, 5 μ m) C18 column, 1 ml/min flow rate and 304 nm as wavelength of analysis. The capecitabine has shown well resolved peak at retention time 6.4 minute (Figure 2).

The developed method was applied for quantification of capecitabine in marketed formulation. The results of assay of capecitabine tables were found to within limits as shown in Table 2. The capecitabine tablets were found to contain NLT 97% and NMT 103% capecitabine. The developed method was validated using ICH Q2B guidelines. Nearly all validation parameters were studied for developed method. The capecitabine was found linear in concentration range of 50-150 μ g/ml with correlation coefficient value, 0.9977. The accuracy study was carried out by recovery study whereby recovering pure capecitabine from tablet solution. The precision study was carried out by analyzing tablet solution with change in time of analysis (inter day and intraday). The robustness was studied by flow rate and temperature of analysis variations. The specificity study was carried out by stability study i.e. by forced degradation of capecitabine in acidic, alkaline and oxidative conditions. The retention time of capecitabine was not changed in presence of its degradation products, confirming specificity of method. The system suitability parameters like tailing factor, therotical plates, retention time etc. were found within limits, thereby confirming least influence of instrumental error contributing factors. All the results of analysis and validation are summarized in Table 8.

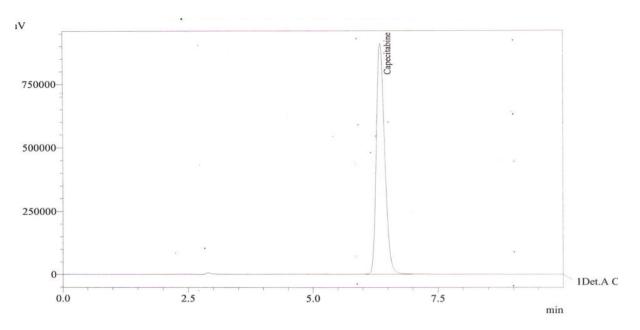


Figure 2. Chromatogram of Standard Drug Retention time for Capecitabine standard is at 6.4 minute.

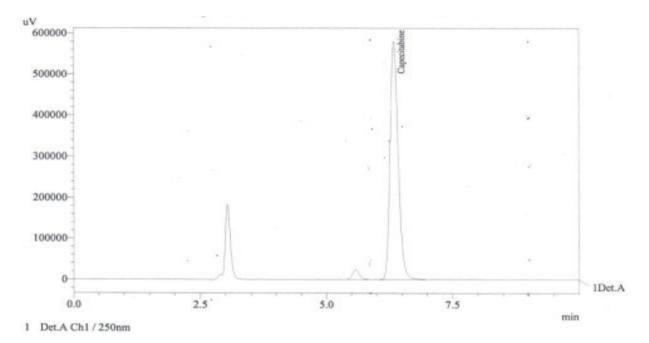


Figure 3. Chromatogram for specificity in 0.1M HCl Peak for Degredant is found at 3.3 minute and for Capecitabine it is at 6.38 minute.

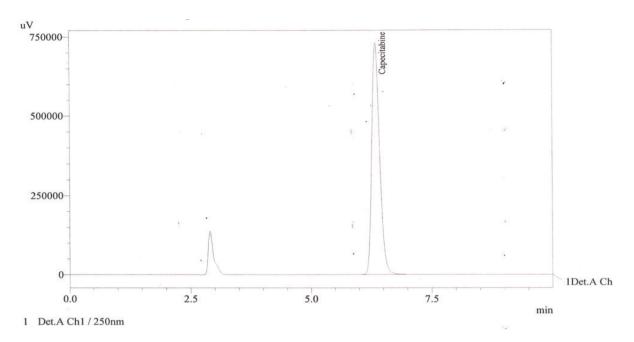


Figure 4. Chromatogram for specificity in 0.1M NaOH Peak for Degredant is found at 2.95 minute and for Capecitabine it is at 6.38 minute.

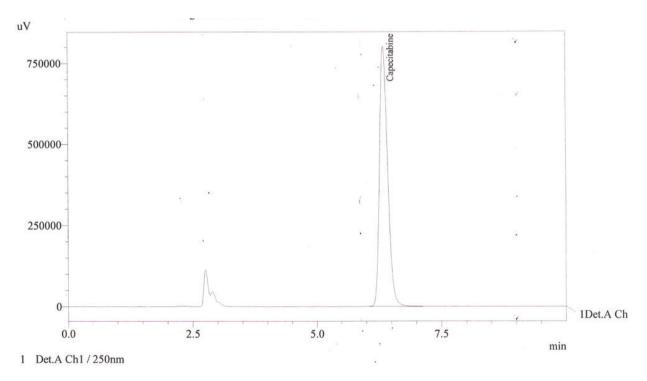


Figure 5. Chromatogram for specificity in 10% H_2O_2 Peak for Degredant is found at 2.7 minute and for Capecitabine it is at 6.38 minute.

CONCLUSION

From results of analysis and validation, it can be concluded that developed method is simple, accurate, precise, sensitive, and selective for quantification of capecitabine from its marketed formulations and it can be used for pharmacokinetic and biopharmaceutical studies with slight modification in optimized chromatographic conditions.

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